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Chemistry (4th Edition) Burdge, Julia Publisher McGraw-Hill Publishing Company ISBN 978-0-07802-152-7

Section 6.2: Factors Affecting Reaction Rates Section 6.2 ...

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Chapter 6 Review, pages 396–401

in 6-sided patterns; other imagined, explanatory properties. 9. A nonpolar solvent such as mineral spirits (Varsol or turpentine) should dissolve grease, according to the “like dissolves like” rule, because grease is a nonpolar substance. 10. (a) C 6 H 12 O 6(s), because it is polar. O HO H O — C — O H H H H H H H

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6H 12O 6(s) is: $6 \text{C(s)} + 6 \text{H}_2\text{(g)} + 3 \text{O}_2\text{(g)} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6\text{(s)}$ 28. A thermochemical equation for the formation of aluminum oxide, $\text{Al}_2\text{O}_3\text{(s)}$ with the energy term separate is: $2 \text{Al(s)} + 3 \text{O}_2\text{(g)} \rightarrow \text{Al}_2\text{O}_3\text{(s)} \Delta H_f^\circ = -1676 \text{ kJ/mol}$ A thermochemical equation for the formation of aluminum oxide, $\text{Al}_2\text{O}_3\text{(s)}$ with the energy term embedded in the ...

6.1 DEFINING A SOLUTION

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Section 6.5 Questions. p. 400 Question 5 Enzymes have application in both body chemistry and industrial reactions. Write a brief report describing the function of an enzyme (not yet mentioned in this chapter) in the human body or in industry. Enzymes and Enzyme Activity This private website is managed by Ralph Logan.

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Unit 6 Chemical Energy - Nelson

Answers to Chapter 6 Review Questions (Student textbook pages 392-7) 1. d 2. b

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3. c 4. a 5. e 6. d 7. d 8. b 9. d 10. e 11. a 12. d 13. c 14. d 15. Rate of reaction is equal to the change in the quantity of a reactant or a product divided by the time during which the change occurs. 16. Sample answer s: a. mL/s b. kPa/s c. pH/s 17. a.

Textbook Solutions Chem 11 Nelson chapters 1-6 - orianaufa

12 O 6 (aq) 6 O 2 (g) ... Nelson Web site, in the Chemistry Review unit, and in the Appendices. A Unit Pre-Test is also available online. (a) (b) (c) (d) Figure 1 ... the chapter. 1. Consider the following changes: photosynthesis, cellular respiration, and combustion of gasoline. Classify each of these changes as exothermic or endothermic.

Nelson Education - Secondary Science - Chemistry 12

Copyright © 2012 Nelson Education Ltd. Chapter 6: Chemical Kinetics 6.2-2 (b) Answers may vary. Sample answer: Costs and Benefits of Use of Butylated Hydroxytoluene (BHT) in the Food Industry Cost Benefit • Since BHT is broken down and absorbed by the body, it has biological effects, some of which may be harmful.

Chapter 5 Review, pages 338-339

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Copyright © 2012 Nelson Education Ltd. Chapter 6: Chemical Kinetics 6-6 28. For the equation $\text{Ag}^+(\text{aq}) + \text{Cl}^-(\text{aq}) \rightarrow \text{AgCl}(\text{s})$, the rate law will be $\text{rate} = k[\text{Ag}^+(\text{aq})][\text{Cl}^-(\text{aq})]$ because the reaction has a single step involving two reactants. 29. Reactions of substances containing covalent bonds take longer than reactions of ions

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