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Also, in formal writing we should use the name of the compound rather than its symbols. Therefore we need to learn how to say the proper names of ionic substances. Model 1 Cations with one charge NaCl - Sodium chloride CaS - Calcium sulfide Ag<sub>2</sub>S - Silver sulfide Zn<sub>3</sub>P<sub>2</sub> - Zinc phosphide Al<sub>2</sub>O<sub>3</sub> - Aluminum oxide DyCl<sub>3</sub> - Dysprosium chloride 1.

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Examine the traditional ion names shown below for selected metals. Write a rule for using the “-ous” and “-ic” endings for metal ions. 2+Cu cupric Fe<sup>3+</sup>ferric Sn<sup>4+</sup>stannic Pb<sup>4+</sup>plumbic Cu<sup>1+</sup>cuprous Fe<sup>2+</sup>ferrous Sn<sup>2+</sup>stannous Pb<sup>2+</sup>plumbous. 27. Complete the table and write the traditional name for each ionic compound.

**Mrs. Primatesta**

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best we can do is to supply students with the chance to learn. You may use any mixture of the ions utilized within this activity, so long as new compounds are named.

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Naming ionic compound with polyvalent ion. Worked example: Finding the formula of an ionic compound. Practice: Predict the charge on monatomic ions. Practice: Naming ionic compounds. This is the currently selected item. Practice: Find the formula for ionic compounds.

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Al is a metal, so the elements are held together by ionic bond. No prefixes are used when naming ionic compounds. Boron is a metalloid/non-metal, so forms covalent bonds with non-metals to make a molecule. Covalent molecules use prefixes in their naming system to indicate the number of atoms of each element is contained in the molecule.

### 18 Naming Molecular Compounds-S

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3. Consider the ions of iron (Fe) and sulfur (S). Write chemical formulas for all possible ionic compounds involving these ions, using the simplest ratio(s) of iron (Fe) and sulfur (S). Keep in mind that the sum of the charges in an ionic compound must equal zero. 115 STOP FeS and S Naming Ionic Compounds

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Ternary ionic compounds have three or more elements involved, binary compounds only two. 18. Consider the compound iron(III) hydroxide in Model 3. a. How many hydroxide ions ( $\text{OH}^-$ ) are combined with an iron(III) ion ( $\text{Fe}^{3+}$ )? Three hydroxide ions. b. Is your answer to part a the only combination of iron(III) and hydroxide that should exist in nature?

### **Naming Ionic Compounds Pogil Answer**

Predicting and Naming Polyatomic Ionic Compo Worksheet Compound Formula Element or Name of Compound PO atomic Cation Aluminum Carbonate Aluminum Acetate Aluminum Hydroxide Name Element or PO atomic Anion a/J02 Given the following polyatomic ionic compounds, fill in the name of the compound from its formula. Compound Formula Compound Name  $\text{BaCO}_3$   $\text{NaOH}$   $\text{NH}_4\text{Cl}$   $\text{CdSO}_4$   $\text{Ag}_2\text{CO}_3$   $\text{KC}_2\text{H}_3\text{O}_2$ .

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