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The total positive charge of the cations in the compound equals the total negative charge of the anions in the compound. Describe the energy change associated with ionic bond formation, and relate it to stability. Ionic bond formation is

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exothermic; the lower energy product is more stable than the original reactants.

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Answers could include reddish- yellow color, conductor of heat and electricity, malleable, melting point of 1084°C , and boiling point of 2562°C .

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