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13. All acid—base reactions have two conjugate acid—base pairs. One conjugate acid—base pair in the reaction in Model 3 is $\text{H}_2\text{O}/\text{H}_3\text{O}^+$. List the other acid—base pair in the reaction. 14. Why is HCO_3^- considered the "acid" part of the pair in the reaction in Model 3? 15. Is CO_3^{2-} considered the "base" part of the pair in the reaction in Model 3? 16.

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STOP Conjugate Acid $\text{CO}_2(\text{aq}) + \text{H}_2\text{O}(\text{aq})$ Model 3 — Conjugate Acid—Base Pairs
Base + 1-120 (l) $\text{HCO}_3^-(\text{aq})$ 13. All acid—base reactions have two conjugate acid—base pairs. One conjugate acid—base pair in the reaction in Model 3 is $\text{H}^+/\text{H}_2\text{O}$. List the other acid—base pair in the reaction. 14.

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